TABLE II EFFECT OF INHIBITORS

Inhibitors were mixed with the enzyme before the addition of the other components of the assay as described in Table I and including 0.1 μ mole vitamin K

Compound added	Final concentration	% Inhibition	
KCN	10 ⁻³ M		
a-a' Dipyridyl	6·10 ^{−3} M	67	
o-Phenanthroline	5·10 ⁻³ M	90	
Na diethyldithiocarbamate	$10^{-3} M$	29	
Salicylic acid	$10^{-3} M$	32	
8-Hydroxyquinoline	$10^{-3} M$	29	
Thiourea	$10^{-3} M$	41	
2-Heptyl-4-hydroxyquinoline	4·10 ⁻⁴ M	95	

Work is now in process to clarify this system and its physiological significance. We are grateful to Dr. Sols for valuable suggestions in the preparation of the manuscript.

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- ¹ S. Taniguchi, R. Sato and F. Egami, Symposium on Inorganic Nitrogen Metabolism, Johns Hopkins University Press, Baltimore, 1956, p. 87.
- ² D. J. D. NICHOLAS AND A. NASON, J. Bacteriol., 69 (1955) 580.
- ³ D. H. LOWRY, N. J. ROSEBROUGH, A. L. FARR AND R. L. RANDALL, J. Biol. Chem., 193 (1951) 265.
- ⁴ W. Wosilait and A. Nason, J. Biol. Chem., 208 (1954) 785.
- ⁵ A. MEDINA AND D. J. D. NICHOLAS, Biochim. Biophys. Acta, 23 (1957) 440.

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Biosynthesis of hyaluronic acid by cell-free extracts of group-A streptococci*

Previous studies have demonstrated that both the glucosamine and glucuronic acid moieties of HA^{**} synthesized by Group-A streptococci are derived from glucose¹. The discovery of uridine nucleotides containing N-acetylated amino sugars² and glucuronic acid³ and the established role of uridine nucleotides as glycosyl donors⁴ suggested that these compounds are intermediates in the biosynthesis of mucopolysaccharides. CIFONELLI AND DORFMAN⁵ have demonstrated the presence of UDPAG, UDPGA and other non-identified fractions containing uronic acid and N-acetyl-glucosamine in a strain of Group-A streptococcus grown under conditions optimal for the production of HA. The only direct evidence in support of the role of uridine nucleotides in acid mucopolysaccharide synthesis is given by the reports of Glaser and Brown^{6,7} that extracts of Rous sarcoma incorporate into HA, ¹⁴C from labeled UDPAG, AG-6-P+UTP, or UDPG. However, a large portion of the radioactivity was lost on reprecipitation or electrodialysis. No incorporation was obtained when labeled UDPGA was used.

It is the purpose of this communication to present evidence that a cell-free extract of a strain of Group-A streptococcus (A111, Type 18) incorporates glucoronic acid from UDPGA. This system has an absolute requirement for UDPAG and Mg⁺⁺ and a relative requirement for AG-1-P and ATP.

Cells were grown as previously described⁵, harvested by centrifugation at $30,000 \times g$ for 10 min, and washed twice with 0.05 M phosphate buffer, pH 7.0. Following treatment in a

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^{**} The following abbreviations are used in this paper: hyaluronic acid, HA; uridine diphospho-N-acetylglucosamine, UDPAG; uridine diphosphoglucoronic acid, UDPGA; uridine diphosphoglucose; uridine triphosphate, UTP; adenosine triphosphate, ATP; diphosphopyridine nucleotide, DPN; N-acetylglucosamine-1-phosphate, AG-1-P; N-acetylglucosamine-6-phosphate, AG-6-P.

Raytheon sonic oscillator at 9 kc, the sonicate was centrifuged at 10,000 \times g for 10 min at 4° and the supernatant used. The UDPGA was isolated from another strain of streptococcus (Richards strain) by the method previously reported⁵. Tritiation was carried out by the method of Wilzbach⁸ and repurification by column chromatography using the neutral ammonium acetate–ethanol solvent of Paladini and Leloir⁹. The repurified material showed the same activity as a glucuronyl donor with p-nitrophenol as the highly purified starting material¹⁰. The AG-1-P was prepared by the method of Maley, Maley and Lardy¹¹. UDPAG was obtained from the Sigma Chemical Company.

The HA from each sample was isolated and purified in the following fashion. After the indicated incubation time, 1.07 mg HA (isolated from the growth supernatant of the same organism by the method of CIFONELLI AND MAYEDA¹²) was added and the sample boiled for 5 min. Sufficient additional HA was added to make a total of 10.7 mg. After adjusting the volume to 10 ml the tubes were centrifuged for 20 min at 2000 r.p.m. The supernatant was dialyzed successively for 6 h against running tap water, 10 h against 15 l distilled H2O, and finally for 24 h against 15 l distilled H_2O . The sample was centrifuged for 20 min at 30,000 \times g and to the supernatant were added I drop 4 N NaOH and I drop phenolphthalein. The solution was brought just to the point of boiling, cooled, neutralized with 3-4 drops I N HCl, and passed over an 800-mg column of Norite-A-20% stearic acid. The latter treatment removes all ultraviolet-absorbing material. The HA was precipitated as the cetyl pyridinium complex by the addition of 1-2 drops 1 MNa₂SO₄ and 1.5-2.0 ml 1 % cetyl pyridinium chloride. The complex was centrifuged, washed 3 times with H₂O, and dissolved in 5 ml methanol. Insoluble material was removed by centrifugation. The HA was precipitated from the supernatant with 1 vol. glacial acetic acid and washed 3 times with methanol, once with methanol-ether, twice with ether and dried in a vacuum desiccator over Drierite. The HA was dissolved in 1 ml 1 M hyamine^{13,14} and the radioactivity determined in the Tri-Carb Liquid Scintillation Spectrometer, using 2,5-diphenyloxazole as a phosphor in toluene. HA prepared by this method showed a hexosamine: uronic acid: N molar ratio of 1:0.99:0.9.

The results of a typical experiment are illustrated in Table I. Omission of MgCl₂ or UDPAG resulted in failure of incorporation of radioactivity while omission of AG-I-P or ATP caused a significant diminution.

When the isolated HA was repurified by adsorption and elution from Dowex-1 Cl⁻, no change in specific radioactivity was observed.

In other experiments employing the same components as above, but utilizing AG-1-P as the radioactive component, radioactive HA was obtained. Under these conditions the UDPGA could be replaced by UDPG and DPN. The inclusion of UTP in the presence of AG-1-P abolishes the absolute requirement for UDPAG.

In one experiment utilizing an enzyme obtained from a 30-min sonicate (Table I), 12.5% of uronic acid of the UDPGA was incorporated into HA. This calculation was based on the assumption that the UDPGA was uniformly labeled. Correction for the presence of HA in the enzyme preparation results in an increase in the estimate of incorporation.

TABLE I INCORPORATION OF 3H From 3H -UDPGA into HA

All tubes contained a total volume of 1.5 ml including 0.75 ml of enzyme with a final phosphate concentration (pH 7.0) of 0.05 M. 10.7 mg carrier HA was added to each tube. The ³H-UDPGA used had an activity of 580,000 counts/min/µmole.

Radioactivity counts/min/mg H.	ATP µmole	MgCl ₂ μmoles	N-Ac-Gm-1-P μmole	UDPAG µmole	³H-UDPGA μmole	Time min
o	1.0	10.0	0.1	1.0	1.0	0
820	1.0	10.0	0.1	1.0	1.0	130
610	1.0	10.0	O	1.0	1.0	130
3	1.0	10.0	1.0	О	1.0	130
15	1.0	o	1.0	1.0	1.0	130
600	o	10.0	1.0	1.0	1.0	130
2600	1.0	10.0	0.1	1.0	0.1	130*

^{*}The enzyme used in this preparation was obtained from a 30-min sonicate, while that used in all other test combinations was from a 15-min sonicate.

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- A. Dorfman, Pharmacol. Revs., 7 (1955) 1.
 E. Cabib, L. F. Leloir and C. E. Cardini, J. Biol. Chem., 203 (1953) 1055.
- ³ G. J. DUTTON AND I. D. E. STOREY, Biochem. J., 57 (1954) 275.
- ⁴ L. F. LELOIR AND E. CABIB, J. Am. Chem. Soc., 75 (1953) 5445.
- ⁵ J. A. CIFONELLI AND A. DORFMAN, J. Biol. Chem., 228 (1957) 547.
- ⁶ L. Glaser and D. Brown, Proc. Natl. Acad. Sci. U.S. 41 (1955) 5.
- ⁷ L. Glaser, Thesis, Washington University, 1956.
- ⁸ K. E. WILZBACH, J. Am. Chem. Soc., 79 (1957) 1013.
- ⁹ A. G. PALADINI AND L. F. LELOIR, Biochem. J., 51 (1952) 426.
- 10 K. J. ISSELBACHER, Recent Progr. in Hormone Research, 12 (1956) 134.
- ¹¹ F. Maley, G. F. Maley and H. A. Lardy, J. Am. Chem. Soc., 78 (1956) 5303.
- 12 J. A. CIFONELLI AND M. MAYEDA, Biochim. Biophys. Acta, 24 (1957) 397.
- 13 J. M. PASSMANN, N. S. RADIN AND J. A. D. COOPER, Anal. Chem., 28 (1956) 484. 14 F. EISENBERG, JR., Northwestern University Conference on Liquid Scintillation Counting, Aug. 20-22, 1957, Evanston, Illinois.

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The role of deoxyribonucleic acid in 32P incorporation into isolated nuclei *

The importance of deoxyribonucleic acid (DNA) in the synthesis of nuclear protein and ribonucleic acid (RNA) has been demonstrated, but its action does not seem to be very specific1,2. Since different workers have observed the incorporation of various precursors into DNA in isolated nuclei8-5, we have undertaken to clarify the role of pre-existing DNA in DNA synthesis with rabbit appendix nuclei.

Nuclei isolated from the appendix and Peyer's patch of adult rabbits by a slight modification of the method of Allfrey et al. were incubated with 32P-orthophosphate according to Friedkin AND WOOD⁴. The procedure of digestion with pancreatic deoxyribonuclease (DNase) and subsequent treatment of nuclei were essentially similar to the ones employed by Allfrey et al.1. I ml of incubation mixture containing about $3\cdot 10^8$ nuclei and 10 μ C 32 P were shaken at 37° for 90 min. The acid-soluble organic-phosphate fraction (OASP) was obtained according to Hokin and HOKIN⁶. Nucleic acids were extensively purified by the routine method used in this laboratory⁷. Radioactivity incorporated into RNA and DNA was evenly distributed among various nucleotide fractions obtained chromatographically (Dowex-I, formic acid systems) following alkaline hydrol-

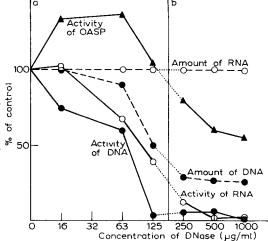


Fig. 1. The effect of DNase treatment of isolated nuclei. Activity refers to the total With 5 radioactivity (counts/min/tube). nucleic acids, this was calculated from the $\, \stackrel{\ \, _\circ}{\sim} \,$ specific activity (counts/min/ μg P) of purified samples and the total amount (µg P/tube) determined by the conventional Schneider method. Specific activities of OASP, RNA and DNA of untreated control nuclei were 1076, 32, 2.4 (Expt. a) and 759, 43. 0.8 (Expt. b).

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